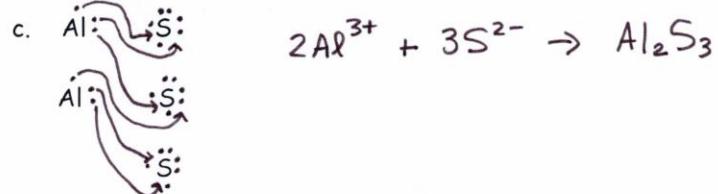
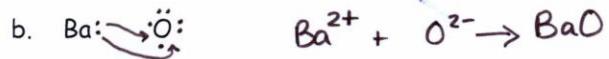
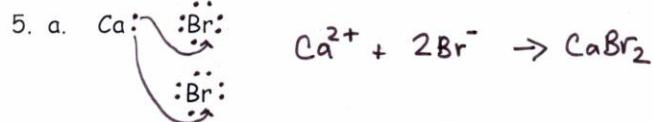


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8. The electronegativity of a bonded atom is its attraction for electrons in the bond.

The electron affinity is the energy involved when a gaseous atom gains an electron.

10. a. F b. Br c. Cl d. N

11. a. Ionic b. Polar Covalent c. Ionic d. Polar Covalent
e. Ionic f. Ionic g. Nonpolar Covalent
h. Nonpolar Covalent i. Polar Covalent