Homework Answers
Hill-Petrucci Chapter 3 & 12
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## #86.

- a.  $0.500 \text{ M} \text{ H}_2\text{S}O_4$
- b. 5.71 M C<sub>2</sub>H<sub>5</sub>OH
- c. 6.324 M KOH
- d. 0.0364 M H<sub>2</sub>C<sub>2</sub>O<sub>4</sub>
- e.  $0.07770 \text{ M } C_6H_{14}O_4$
- f. 0.783 M C<sub>3</sub>H<sub>9</sub>N

## #24.

- a. 3.96 % by mass
- b. 7.31% by mass
- c. 8.21 % by mass
- 48. a. 3.0  $m \text{ Li}_2 SO_4$ b. 0.062  $g \text{ Pb}(NO_3)_2$  can be added
- 49. The solution could be left open to air so that some of the solvent evaporates. When enough solvent has evaporated, the solute will begin to crystallize.